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Answer Key 1. What  
is organic chemistry?

The branch of  
chemistry dealing with

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Chemistry of carbon.

2. How many carbon compounds exist?

Millions 3. How many bonds can carbon form? Four bonds 4.

When was nylon first developed? Why was its development so essential? Before

World War II. A substitute for silk. 5.

What was the first organic synthesis?

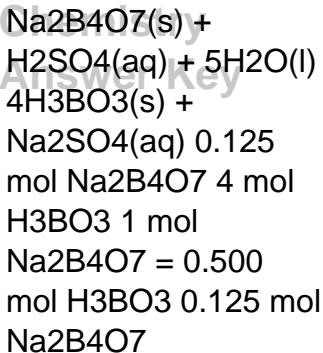
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 $2\text{KCl}(\text{aq}) + \text{I}_2(\text{s})$  0.125  
mol KI 2 mol KCl 2  
mol KI = 0.125 mol  
KCl 0.125 mol KI 1  
mol  $\text{I}_2$  2 mol KI =  
0.0625 mol  $\text{I}_2$  c.

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Chapter 9 Chemical  
Quantities  
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mass of 1 mol Ba =  
137.3 g = 137.3 g

mass of 2 mol Cl = 2  
(35.45 g) = 70.90 g

mass of 8 mol O = 8  
(16.00 g) = 128.0 g

molar mass of  
Ba(ClO<sub>4</sub>)<sub>2</sub> = 336.2 g

b. mass of 1 mol Mg =  
24.31 g = 24.31 g

mass of 1 mol S =  
32.07 g = 32.07 g

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## Chapter 6 Chemical Composition

Answer Key 1. Why is it important to use the correct amount of materials in a chemical reaction? If too little is used the reaction may not proceed very far. The use of too much chemical may result in waste. 2. What names



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are given to the materials at the beginning and end of a chemical reaction? Reactants and products. 3.

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together by chemical bonds. Carbon dioxide and water are two examples, but answers will vary. 3. Rutherford's experiment involved shooting a beam of particles at a thin sheet of metal foil. According to the then-current "plum pudding" model of the atom, most of these

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positively charged particles should have passed right through the foil.

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matter, the  
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