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Coulomb ' s law of
electrostatic attraction
we know that the PE
between two charges Q_1
and Q_2 separated by a
distance r is given by

$$PE = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r}$$

First we consider the
interaction between
 Na^+ ion and $6Cl^-$ ions
at distance r Applying
Coulomb ' s law we

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have $Q_1 Q_2 = (-6e)(+e)$

$$PE_1 = \frac{-6e^2}{4\pi\epsilon_0 r} = -4 \frac{e^2}{4\pi\epsilon_0 r}$$

$$\text{or } 4 \frac{e^2}{4\pi\epsilon_0 r} \text{ or}$$

Similarly, we now

consider 12 Na^+ ions as
second nearest

neighbors at a distance r

$$Q_1 Q_2 = (+12e)(+e) = 12e^2$$

$$PE_2 = \frac{12e^2}{4\pi\epsilon_0 r} = 4 \frac{e^2}{4\pi\epsilon_0 r}$$

$$\text{or } 4 \frac{e^2}{4\pi\epsilon_0 r} \text{ or and}$$

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