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Hydrogen, synthesis gases and their derivatives. 2. Sources of olefinic and aromatic hydrocarbons. 3. The treatment of olefinic C4 and C5 cuts. 4. The treatment of aromatic gasolines.

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#### <u>Petrochemical Processes</u> <u>Volume 1 - Better World</u> Books

Chemistry and Technology of Basic Organic and Petrochemical Synthesis. Volume 1 and Volume 2. N. N. Lebedev, MIR Publishers, Moscow, 1984,333 and 305 pp.

Chemistry and Technology of Page 9/20

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Important Industrial
Dehydrogenation Processes 1.
Preparation of butadiene by
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dehydrogenation of n-butane and n-butenes Annual world production is over 8 mil. t/year 1.1. Two steps process 1. CH 3CH 2CH 2CH 3 = CH 3CH 2CH=CH 2 + H 2, at 560 - 600oC Al 20 3-Cr 20 3 catalyst, in fluidized bed reactor, x butene = 30 - 34 % 2. CH 3CH 2CH=CH 2 ...

Important Petrochemical
Processes - kau
Alkanes are important
because of the large volume
of their use in
petrochemical synthesis. The
lower gaseous hydrocarbons
(methane, ethane, propane,
butane, and the pentanes)
and the liquid or solid
paraffins (from C 6 to C 40)

are used to make various chemical products. The lower paraffin hydrocarbons are isolated from natural and casinghead ...

#### Petrochemical Synthesis | Article about Petrochemical

. . .

The field of petrochemicals started some years ago with the simple addition reaction of water to propylene for the production of isopropyl alcohol. Currently, the petrochemical industry has become a multi-billion dollar enterprise which encompasses a wide field of chemical products. Almost all the basic organic reactions such as

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hydrogenation, alkylation, substitution, drocarbons polymerization, etc. are ...

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<u>Chemistry of Petrochemical</u> <u>Processes - 2nd Edition</u> <u>Page 14/20</u>

@misc{etde 6778332, title = {Petrochemical processes. Volume 2} author = {Chauvel, A, Lefebvre, G, and Leprince, P} abstractNote = {This book provides a technical and economic analysis of petrochemical techniques. Are examined: ethylene oxide, propylene oxide, glycol, acetic derivatives, alcohols (synthesis of ethanol, isopropanol, butanols and higher alcohols), phenol, ketones, monomers ...

Petrochemical processes. Volume 2 (Book) | ETDEWEB Hydrocarbon Processing -Petrochemical Processes 2001, Gulf Publishing

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Company, 2001. Hydrocarbons
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Publishing Company, 2003.
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2005, Gulf Publishing
Company, 2005. Hydrocarbon
Processing - Petrochemical
Processes 2010, Gulf
Publishing Company, 2010.

Bibliografía y Referencias - UTN FRBA DIQ Proyecto
In methanol synthesis from synthesis gas (composition, vol %: H2 76.6, C02 19.8, and N2 3.6), high C02 conversion of 84–99.6% is achieved at 240–260°C and a pressure of 5.3 MPa with low selectivity in regard to the

side reaction (<4.7% atives synthesis of CO). The maximum specific productivity of methanol at 260°C is 1.24 kg (kgcat h)-1.

Methanol and Dimethyl Ether Synthesis from CO 2 and H 2 in ...

Ammonia synthesis, produced on a > 150 Mt world scale in a multistep process starting mainly from natural gas, is the largest-scale chemical process and top largest chemical process in terms of energy consumption, with about 2.5 EJ use on a world scale, and between 1.8–2.1 ton CO 2 emitted per ton of NH 3 produced, depending on

plant efficiency erivatives

And Major Hydrocarbons

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Science 06 Mar 2020: Vol. 367, Issue 6482, eabb0992 DOI: 10.1126/science.abb0992

Erratum for the Review:

"What would it take for renewably ...

Succinate is a valuable platform chemical for multiple applications.

Confronted with the exhaustion of fossil energy resources, fermentative succinate production from renewable biomass to replace the traditional petrochemical process is

Page 19/20

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