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 $= M_2V_2$  ] ||

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own symbol, which  
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mol/L would be  
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value of 5 molar.

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where the  
subscripts "1" and  
"2" refer to the  
solution before and  
after the dilution,  
respectively. Since  
the dilution process  
does not change  
the amount of  
solute in the  
solution,  $n_1 = n_2$ .  
Thus, these two



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Equations may be set equal to one another:

$$M_1V_1 = M_2V_2$$

This relation is commonly referred to as the dilution equation.

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calculator tool  
calculates the  
volume of stock  
concentrate to add  
to achieve a  
specified volume  
and concentration.  
The calculator uses  
the formula  $M_1 V_1 = M_2 V_2$  where  
"1" represents the



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concentrated  
conditions (i.e.  
stock solution  
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volume) and "2"  
represents the  
diluted ...

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