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Pradeep Singh elasticity A

*bottle has an opening of*

*radius  $a$  and length  $b$  A cork*

*of length  $b$  and radius  $a + \frac{a}{2}$*

*wher*

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Modern Chemistry 71 Chapter  
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Test B, continued 17. A  
substance combines with  
oxygen, releasing a large

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amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 24 Chapter Test . Name Class Date I  
Chapter Test B, continued 7.  
The discovery of the nucleus was a result of Rutherford's observation that a small percentage of the positively charged particles bombarding the metal's surface a. were slightly deflected as they passed through the metal.

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Modern Chemistry 1 Chapter  
Test Assessment Chapter Test  
B Teacher Notes and Answers  
2 Measurements and  
Calculations TEST B 1. a 2.  
c 3. c 4. a 5. c 6. c 7. a  
8. a 9. time 10. mass 11.  
density 12. energy or work  
13. length or distance 14.  
volume 15. area 16.  
qualitative 17. quantitative  
18. qualitative 19.  
quantitative 20.  $3.00 \pm 10^5$   
km/s 21. three 22.  $2.6 \pm 10^2$   
g 23.  $2.50 \pm 10^2$  L  
24. quantity 25.

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Modern Chemistry 141 Chapter  
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Test B, continued 18. When a  
strong acid is titrated with  
a weak base, the pH of the  
solution at the equivalence  
point is than 7. 19. A is a  
highly purified solid used  
to check the concentration  
of a standard solution. 20.  
A 1 M solution of NaOH will  
have a pH that is

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Chemistry Chapter Test  
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Teacher Notes and Answers 5  
The Periodic Law TEST B 1. a  
2. c 3. d 4. d 5. a 6. a 7.  
c 8. a 9. lanthanides 10. 2

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11. fourth 12. transition  
elements 13. 32 14. valence  
electrons 15. electron  
affinity 16.  
electronegativity 17.  
ionization energy 18.  $3s^2$   
 $3p^4$  19. atomic radius 20 ...

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Modern Chemistry 33 Chapter  
Test Name Class Date Chapter  
Test B, continued 15. The  
energy state of an atom is  
called its ground state. 16.  
The number of waves that  
pass a point in one second  
is called. 17. When an  
electron drops from a higher-  
energy state to a lower-  
energy state, a(n) spectrum  
is produced. 18.

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Modern Chemistry 91 Chapter  
Test Name Class Date Chapter  
Test B, continued 27.  
Distinguish between ionic  
crystals and metallic  
crystals. PART V Write the  
answers to the following  
questions on the line to the  
left, and show your work in  
the space provided. Modern  
Chemistry Chapter 12 Review  
Answer Key

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Modern Chemistry 20 Chapter  
Test Name Class Date Chapter  
Test A, continued \_\_\_\_\_13.

To calculate the number of  
atoms present in 2.0 mol of  
an element, you would a. add  
Avogadro's number of atoms  
per mole to 2.0 mole. b.  
subtract Avogadro's number  
of atoms per mole from 2.0  
mole. c.

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Chemistry 3 Chapter Test  
Chapter Test B, continued 16  
Modern chemistry chapter 3  
test b answers. The measure  
of the ability of an atom in  
a chemical compound to  
attract electrons from  
another atom in the compound  
is called \_\_\_\_\_. 17. The  
energy required to remove  
one electron from an atom is  
called its \_\_\_\_\_. 18.

## ~~Modern Chemistry Chapter 3 Test B Answers~~

6 Lessons in Chapter 13:  
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Chemistry Chapter 13: Ions  
in Aqueous Solutions and  
Colligative Properties  
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your knowledge with a  
30-question chapter ...

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Chemistry Chapter 13: Ions  
in Aqueous ...~~

CHAPTER 3 TEST continued

Date Class FILL IN THE BLANK

Write the correct term (or  
terms) in the space

provided. 9, If a particular  
ompound is composed of  
elements A and B, the ratio  
of the mass of B to the mass  
of A will alw ys be the  
same. This is a statement of  
the law of exactly 12 g of  
carbon-12 is referred to as  
a(n) 11.

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(b) positive ions. (d)

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negative ions. 3. b Compared with the neutral atoms involved in the formation of an ionic compound, the crystal lattice that results is (a) higher in potential energy. (c) equal in potential energy. (b) lower in potential energy. (d) unstable. 4. b The lattice energy of compound A is greater in magnitude than that of ...

## ~~6 Chemical Bonding~~

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b. Explain how an atom can exist in this state. Atoms consist of a positively charged nucleus, made up of protons and neutrons, that is surrounded by a negatively charged electron cloud. The positive and negative charges combine to form a net neutral charge.

MODERN CHEMISTRY ATOMS: THE  
BUILDING BLOCKS OF MATTER 19

~~3 Atoms: The Building Blocks  
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Modern Chemistry 137 Chapter  
Test Name Class Date Chapter  
Test A, continued \_\_\_\_\_19.



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In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given.

\_\_\_\_\_20. In the figure on the previous page, the volume of titration standard

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CHAPTER 22 TEST Nuclear  
Chemistry Class MULTIPLE  
CHOICE On the line at the  
left of each statement,

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Write the letter of the choice that best completes the statement or answers the question. After converting units, the nuclear mass defect is equivalent to the

- atomic mass
- electrostatic force
- energy of chemical reaction

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CHAPTER 16 REVIEW Reaction

Energy SECTION 2 SHORT

ANSWER Answer the following

questions in the space

provided. 1. For the

following examples, state

whether the change in

entropy favors the forward

or reverse reaction: forward

reaction a.  $\text{HCl}(l) \rightleftharpoons \text{HCl}(g)$

reverse reaction b.  $\text{C}_6\text{H}_{12}\text{O}_6$

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6(aq) ?? C 6H 12O 6(s)

forward reaction c. 2NH 3(g)

...

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