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Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the\_\_ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a\_\_ noble gas. Ions that are formed from more than one atom are called\_\_

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: True a. If the formula mass of one molecule is  $x$  amu, the molar mass is  $x$  g/mol. False b. Samples of equal numbers of moles of two different chemicals

### *7 Chemical Formulas and Chemical Compounds*

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Quiz Modern Chemistry 33 . Name Section Quiz, continued Class Date 6. The Lewis structure of which of the following compounds does not follow the octet rule? a. NH<sub>3</sub> b. BF<sub>3</sub> c. d. H<sub>2</sub>O 7. In a double covalent bond, a. one atom has more than eight valence electrons.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

### *5 The Periodic Law*

Modern Chemistry 137 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. \_\_\_\_20. In the figure on the previous page, the volume of titration standard

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