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Chapter 3 Empirical Formulas, Stoichiometry, and Hydrates AP Chem Tomei

Zumdahl Chemistry 7th ed. Chapter 3
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AP Chemistry-Chapter 3 MC Practice Questions Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. Combustion analysis of 0.800 grams of an unknown hydrocarbon yields 2.613 g CO₂ and 0.778 g H₂O. What is the percent composition of the

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AP Chemistry - Practice Test: Chapter 3 Name _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Silver

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has two naturally occurring isotopes with the following isotopic masses: 107.47 Ar 106.90509 108.9047
The average atomic mass of silver is 107.8682 amu .

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____ 1. Combustion analysis of 0.800 grams of an unknown hydrocarbon yields 2.613 g CO_2 and $0.778 \text{ g H}_2\text{O}$. What is the percent composition of the hydrocarbon? a.

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1. In 1928, 3.3 g of a new element was isolated from 660 kg of the ore molybdenite. The percent by mass of this element in the ore was: a. 5% b. 6.6% c. 3.3% d. $5.0 \times 10^{-4} \%$ e. 2.2%
2. A quantitative observation a. contains a number and a unit. b. does not contain a number.

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CHAPTER 3 - Problem 8 The first ionization energy for a neutral atom of chlorine is 1.25 MJ / mol and the first ionization energy for a neutral atom of argon is 1.52 MJ / mol How would the first ionization energy value for a neutral atom of potassium compare to those values?

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