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Then the manometer rule gives  $p_A + \rho g h_{AB} + \rho g h_{BC} - \rho g h_{CD} = p_D$  Here,  $p_A = p_D = 0$ , since points A and D are exposed to the atmosphere.  $0 + 1900 \text{ kg} > \text{m}^3 \cdot 21 \text{g} \cdot 210.6 \text{ m}^2 + 11000 \text{ kg} > \text{m}^3 \cdot 21 \text{g} \cdot 210.8 \text{ m} - h \dots$

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Fluid Mechanics, 6th Ed. Kundu, Cohen, and Dowling Exercise 1.3. The Maxwell probability distribution,  $f(v) = f(v_1, v_2, v_3)$ , of molecular velocities in a gas flow at a point in space with average velocity  $u$  is given by (1.1). a) Verify that  $u$  is the average molecular velocity, and determine the standard deviations ( $\sigma_1, \sigma_2, \sigma_3$ ).

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Solution:  $p_{\text{abs}} = 140 \text{ kPa} \cdot 1.470 \text{ psi} \cdot 101.3 \text{ kPa} = 20.32 \text{ psia}$   $p_{\text{gage}} = p_{\text{abs}} - p_{\text{atm}} = (20.32 \text{ psia}) - (14.70 \text{ psia}) = 5.62 \text{ psi}$   $p_{\text{gage}} = 5.62 \text{ psig}$  c.) Situation: Pressure values need to be converted. Find: Calculate the absolute pressure (psia) corresponding to a pressure of 0.55 bar (gage). Properties:  $p_{\text{atm}} = 14.70 \text{ psi}$ . Solution:  $p_{\text{gage}} = 0.55 \text{ bar} \cdot 1.470 \text{ psi} = 0.8085 \text{ bar} \cdot 1.470 \text{ psi} = 1.1885 \text{ psi}$

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