

# Read Online 13 4 Practice Problems Chemistry

## Answers 13 4 Practice Problems Chemistry Answers

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Problems 1. What volume

would be occupied by 100. g

of oxygen gas at a

pressure of 1.50 atm and a

temperature of 250C?  $m = \frac{PV}{RT}$  (IF

$T = 273K$ ) 2. An air-filled

balloon has a volume of 5 L

at 0.94 atm and 250C. Soon

after, the pressure changes

to 0.99 atm and the temperature

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~~Answers~~ mm Hg. If the mercury level on the side open to the atmosphere is 3.6 mm higher than on the side attached to the cabinet, what is the pressure inside the cabinet in units of kPa? The U-

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~~13-4 Practice Form K The~~  
Sine Function Use the graph at the right to find or estimate the value of  $y = 5 \sin u$  for each value of  $u$ .  
1.  $\pi$  radians 2. 2 radians 3.  $3\pi/2$  radians 4. 4 radians  
Determine the number of cycles each sine function has in the interval from 0 to  $2\pi$ . Then find the period of each function. 5.

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Popular The mole fraction in  
Figure 13.4.4 is that of the  
dissolved solute, and so  
there is a limit. As we



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Answers shall see in the next section, if the solute is volatile, which usually describes miscible liquids, you can have all mole fraction

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Answers Chapter 13 26  
Prentice-Hall, Inc. 13-4  
Practice Problems 1. What volume would be occupied by 100-g of oxygen gas at a pressure of 1.50-atm and a temperature of 25 °C? 6. How many grams of argon would it take to fill a light bulb with a volume of 0.475-L at STP?

~~13-2 Practice Problems - Mr. Steffin's Science Page~~  
13.1 The Nature of Gases - Chemistry & You; 13.1 The Nature of Gases - Sample Problem 13.1; 13.1 The Nature of Gases - 13.1 Lesson Check; 13.2 The Nature of Liquids - Chemistry & You; 13.2 The

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Lesson Check; 13.3 The  
Nature of Solids - Chemistry  
& You; 13.3 The Nature of  
Solids - 13.3 Lesson Check;  
13.4 Changes of State -  
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Figure 13.4.4 is that of the dissolved solute, and so there is a limit. As we shall see in the next section, if the solute is volatile, which usually describes miscible liquids, you

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